

Preparation of chiral propargylic alcohols from α , β -unsaturated esters

Jiong Chun, Hoe-Sup Byun and Robert Bittman*

Department of Chemistry and Biochemistry, Queens College of The City University of New York, Flushing, NY 11367-1597, USA

Received 14 August 2002; revised 6 September 2002; accepted 12 September 2002

Abstract—A series of chiral propargylic alcohols with high enantiomeric excess was prepared by asymmetric dihydroxylation of α , β -unsaturated esters, conversion of the diols to 4-(chloromethyl)-1,3-dioxolane intermediates, and base-induced elimination. © 2002 Elsevier Science Ltd. All rights reserved.

Chiral propargylic alcohols are widely used synthons for many natural products.¹ Among the methods for preparing chiral propargylic alcohols are addition of organometallic reagents to α,β -acetylenic aldehydes in the presence of chiral ligands,² enantioselective aldol additions to α,β -ynals,³ asymmetric reduction of α,β ynones,⁴ enzymatic resolution,⁵ base-mediated double elimination of substrates [1-chloro-2,3-epoxy⁶ and 4-(chloromethyl)-1,3-dioxolane⁷], elimination of chiral vinyl sulfoxides,⁸ alkynylation of aldehydes catalyzed by a chiral ligand,⁹ and reductive cleavage of α,β alkynyl acetals.¹⁰ However, some of these methods provide low enantiomeric excess (ee), especially for the synthesis of long-chain propargylic alcohols. For example, Alpine-Borane reduction of a long-chain ynone gave the *E* analog of **1a** in only 60-65% ee.¹¹ Furthermore, when we used ynone reduction (LiAlH₄ in the presence of Darvon alcohol^{4a} or (R)-Alpine-Borane^{4b}) to prepare 1c (an important synthon for ceramide synthesis), an unsatisfactory optical purity (<80% ee) was obtained. Since it is well known that α,β -unsaturated esters are excellent substrates for the asymmetric dihydroxylation (AD) reaction (giving high yield and very high ee),¹² we have developed a new method for the preparation of 4-(chloromethyl)-1,3-dioxolane intermediates from α,β -unsaturated esters. These dioxolane intermediates are easily converted to chiral propargylic alcohols via double elimination. The configuration at C-3 of the product is established during the AD reaction and remains intact in this reaction sequence. We demonstrate here the utility of this method by preparing a series of representative propargylic alcohols 1a-e (Scheme 1) in high chiral purity.

The Horner–Wadsworth–Emmons (HWE) reaction of aldehydes with a phosphonate reagent provides α , β -unsaturated esters.¹³ Compounds **2a**,¹⁴ **2c**,¹⁵ and **2d** were prepared via HWE reaction of the aldehydes with (*i*-PrO)₂P(O)CH₂CO₂Et. Compound **2e** was synthesized as previously described.¹⁶

Compound 1a is the Z analog of eicos-(4E)-en-1-yn-3ol. The E isomer, which was isolated from a marine sponge, has been synthesized;^{11,17} it exhibits immunosuppressive and antitumor activities.¹⁸ Analog 1a, which may be useful for establishing structure-function relationships, was obtained in geometrically pure form by the reaction sequence illustrated in Scheme 2. The AD reaction of α,β -unsaturated esters **2b**-e provided diols **3b–e** in high yield (\sim 90%). However, diol **3a** was obtained in low yield under the normal AD reaction conditions. The yield of 3a was markedly improved (to 80%) by supplementation of the reaction mixture with additional osmium reagent (K₂OsO₄·2H₂O) and chiral ligand [(DHQ)₂-PHAL]. After conversion of diol 3a to acetonide ester 4a, reduction with DIBAL-H provided alcohol 5a in 92% yield (two steps from 3a). Reduction of 5a with Lindlar catalyst in MeOH provided cis compound 6a, which was converted to chloride 7a (NCS, Ph_3P).¹⁹ (Z)-Alcohol 1a was obtained by treatment of 7a with *n*-BuLi in the presence of HMPA⁷ (81% yield, 96% ee).²⁰ Application of the method used to prepare alcohol 1a also afforded 1b-e with high ee (Scheme 1).²¹

Keywords: asymmetric dihydroxylation; propargyl secondary alcohols.

^{*} Corresponding author. Tel.: (718) 997-3279; fax: (718) 997-3349; e-mail: robert_bittman@gc.edu

^{0040-4039/02/\$ -} see front matter @ 2002 Elsevier Science Ltd. All rights reserved. PII: S0040-4039(02)01968-8



Scheme 1. Outline of the preparation of chiral alcohol 1 from ester 2.



Scheme 2. Representative example of the conversion of ester 2a to alcohol 1. *Reagents and conditions*: (a) AD-mix- α (more K₂OsO₄·2H₂O and (DHQ)₂-PHAL were added), MeSO₂NH₂, *t*-BuOH/H₂O 1:1, rt; (b) Me₂C(OMe)₂, *p*-TsOH, CH₂Cl₂, rt; (c) DIBAL-H, THF, 0°C-rt; (d) H₂, Pd-CaCO₃ (cat.), MeOH, rt; (e) NCS, Ph₃P, CH₂Cl₂, 0°C-rt; (f) *n*-BuLi, HMPA, THF, -42°C-rt.

In summary, we have developed an efficient method that affords both R and S propargylic alcohols in high ee. AD reaction of an α,β -unsaturated ester with either AD-mix α or β , followed by acetonide formation, DIBAL-H reduction, conversion of the resultant primary alcohol to 4-(chloromethyl)-1,3-dioxolane, and elimination with *n*-BuLi in the presence of HMPA gave chiral propargylic alcohols **1a–e** in high ee.

Acknowledgements

This work was supported by National Institutes of Health Grant HL 16660.

References

 Modern Acetylene Chemistry; Stang, P. J.; Diederich, F., Eds.; VCH: Weinheim, 1995.

- See, for example: BouzBouz, S.; Pradaux, F.; Cossy, J.; Ferroud, C.; Falguières, A. *Tetrahedron Lett.* 2000, 41, 8877–8880.
- (a) Niwa, S.; Soai, K. J. Chem. Soc., Perkin Trans. 1 1990, 937–943; (b) Singer, R. A.; Shepard, M. S.; Carreira, E. M. Tetrahedron 1998, 54, 7025–7032.
- (a) Marshall, J. A.; Wang, X.-J. J. Org. Chem. 1991, 56, 4913–4918; (b) Midland, M. M.; Graham, R. S. Org. Synth. 1985, 63, 57–65.
- (a) Takano, S.; Setoh, M.; Yamada, O.; Ogasawara, K. Synthesis 1993, 1253–1256; (b) Waldinger, C.; Schneider, M.; Botta, M.; Corelli, F.; Summa, V. Tetrahedron: Asymmetry 1996, 7, 1485–1488.
- Takano, S.; Samizu, K.; Sugihara, T.; Ogasawara, K. J. Chem. Soc., Chem. Commun. 1989, 1344–1345.
- (a) Yadav, J. S.; Chander, M. C.; Joshi, B. V. *Tetrahedron Lett.* **1988**, *29*, 2737–2740; (b) Takano, S.; Sugihara, T.; Ogasawara, K. *Heterocycles* **1990**, *31*, 1721–1725; (c) Takano, S.; Yoshimitsu, T.; Ogasawara, K. *Synlett* **1994**, 119–120. The generation of 4-(chloro-

methyl)-1,3-dioxolanes via the AD reaction of allylic chlorides and subsequent protection of the diol has limitations, e.g. allylic chlorides are easily hydrolyzed and undergo the AD reaction with lower yield and lower % ee than α , β -unsaturated esters.¹².

- Nakamura, S.; Kusuda, S.; Kawamura, K.; Toru, T. J. Org. Chem. 2002, 67, 640–647.
- Recent examples: (a) Frantz, D. E.; Fässler, R.; Tomooka, C. S.; Carreira, E. M. Acc. Chem. Res. 2000, 33, 373–381; (b) Jiang, B.; Chen, Z.; Xiong, W. J. Chem. Soc., Chem. Commun. 2002, 1524–1525.
- (a) Ishihara, K.; Mori, A.; Arai, I.; Yamamoto, H. *Tetrahedron Lett.* **1986**, *27*, 983–986; (b) Mori, A.; Ishihara, K.; Arai, I.; Yamamoto, H. *Tetrahedron* **1987**, *43*, 755–764.
- Coval, S.; Saucy, G.; Wood, R. D.; Desai, R. C.; Gunawardane, G. P.; Longely, R. E.; Burres, N. US Patent 5166379 (1992); *Chem. Abstr.* **1993**, *118*, 124061v.
- 12. Kolb, H. C.; VanNieuwenhze, M. S.; Sharpless, K. B. Chem. Rev. 1994, 94, 2483–2547.
- 13. For a review of the HWE reaction, see: Maryanoff, B. E. Chem. Rev. 1989, 89, 863–927.
- 14. He, L.; Byun, H.-S.; Bittman, R. J. Org. Chem. 2000, 65, 7627–7633.
- Bonini, C.; Federici, C.; Rossi, L.; Righi, G. J. Org. Chem. 1995, 60, 4803–4812.
- He, L.; Byun, H.-S.; Smith, J.; Wilschut, J.; Bittman, R. J. Am. Chem. Soc. 1999, 121, 3897–3903.
- (a) Sharma, A.; Chattopadhyay, S. *Tetrahedron: Asymmetry* **1998**, *9*, 2635–2639; (b) Lu, W.; Zheng, G.; Cai, J. *Tetrahedron* **1999**, *55*, 4649–4654.
- (a) Gunaseketa, S. P.; Faircolth, G. T. J. Org. Chem. 1990, 55, 6223–6225; (b) Hallock, Y. F.; Cardellina, J. H., II; Balaschak, M. S.; Alexander, M. R.; Prather, T. R.; Shoemaker, R. H.; Boyd, M. R. J. Nat. Prod. 1995, 58, 1801–1807.

- A review of the Mitsunobu reaction: Hughes, D. L. Org. React. 1992, 42, 335–656.
- 20. The % ee was determined by ¹H and ¹⁹F NMR spectroscopy of the Mosher ester obtained by reaction of the alcohol with (S)-(+)-α-methoxy-α-(trifluoromethyl)-phenylacetic acid (MTPA) chloride in the presence of DMAP: Guivisdalsky, P. N.; Bittman, R. J. Org. Chem. 1989, 54, 4637–4642.
- 21. Compound 1a: mp 32.5–33.0°C, $[\alpha]_D^{25}$ +76.6 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 0.88 (t, 3H, J = 6.4 Hz), 1.10–1.60 (m, 26H), 2.03 (s, 1H), 2.11 (m, 2H), 2.49 (d, 1H, J=2.2 Hz), 5.14 (m, 1H), 5.56 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 14.1, 22.7, 27.6, 29.2, 29.3, 29.4, 29.5, 29.6, 27.7, 32.0, 58.0, 72.9, 84.1, 128.7, 134.1; HR-MS [DEI, M^+] m/z calcd for $C_{20}H_{36}O$ 292.2766, found 292.2761. Compound 1b: $[\alpha]_{D}^{25}$ -28.0 (c 1.0, CHCl₃) (lit.⁸ $[\alpha]_{D}^{21}$ -26.7 (c 1.5, CHCl₃); Fluka catalog: $[\alpha]_{D}^{20}$ -28 (c 3.2, CHCl₃)). Compound 1c: mp 40.0-41.0°C; [\alpha]_D²⁵ +2.60 (c 1.0, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 0.88 (t, 3H, J=6.6 Hz), 1.20–1.60 (m, 20H), 1.72 (m, 2H), 2.41 (br s, 1H), 2.44 (d, 1H, J=3.7 Hz), 4.36 (m, 1H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃) δ 14.1, 22.7, 25.1, 29.3, 29.5, 29.62, 29.69, 29.7, 32.0, 37.6, 62.3, 72.8, 85.1; HR-MS [DCI, MH^+] m/z calcd for $C_{15}H_{29}O$ 225.2218, found 225.2211. Compound 1d: mp 46.0-47.0°C; $[\alpha]_D^{25}$ +6.96 (c 2.3, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 0.88 (t, 3H, J=6.6 Hz), 1.20–1.40 (m, 26H), 1.57 (m, 2H), 2.44 (d, 1H, J=2.2 Hz), 2.57 (d, 1H, J=4.8Hz), 3.52 (m, 3H), 3.60 (m, 1H), 4.53 (m, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 14.1, 22.7, 26.0, 29.3, 29.45, 29.53, 29.60, 29.62, 29.69, 29.7, 31.9, 61.4, 71.8, 73.6, 74.0, 81.7; HR-MS [DCI, MNH_4^+] m/z calcd for $C_{20}H_{42}NO_2$ 328.3216, found 328.3216. Compound 1e: mp 50.0-51.0°C (lit.^{5a} mp 52–53°C); $[\alpha]_D^{25}$ +28.6 (*c* 1.0, CHCl₃). The NMR spectra of 1b and 1e were in full accordance with the literature data.